

**STATE UNIVERSITY OF NEW YORK  
COLLEGE OF TECHNOLOGY  
CANTON, NEW YORK**

**COURSE OUTLINE  
CHEM 430-BIOCHEMISTRY**

**Prepared By: Nicole Heldt, Ph.D.**

**SCHOOL OF SCIENCE, HEALTH & CRIMINAL JUSTICE  
May 2015**

## CHEM 430 Biochemistry

- A. TITLE: Biochemistry
- B. COURSE NUMBER: CHEM 430
- C. CREDIT HOURS: 3
- D. WRITING INTENSIVE COURSE (OPTIONAL): No
- E. COURSE LENGTH: 15 weeks
- F. SEMESTER(S) OFFERED: Spring
- G. HOURS OF LECTURE, LABORATORY, RECITATION, TUTORIAL, ACTIVITY:  
3 hours lecture per week.

H. CATALOG DESCRIPTION:

This course provides an introduction to the structure and function of biological macromolecules, bioenergetics, and transfer of genetic information. Emphasis will be on protein structure and function, enzyme catalysis, an overview of energy metabolism, and the maintenance and expression of genetic information.

- I. PRE-REQUISITE: CHEM 302 Organic Chemistry II

- J. STUDENT LEARNING OUTCOMES: Upon completion the student will be able to:

<i>Course Objective</i>	<i>Institutional SLO</i>
1. Describe the molecular basis of life and the organization of macromolecules	2. Crit. Thinking 3. Prof. Competence
2. Describe the physical and chemical properties of proteins and enzymes; relate structure to function of proteins; and explain enzyme catalysis and regulation; and apply thermodynamic and kinetic theories to enzyme reactions	2. Crit. Thinking 3. Prof. Competence
3. Describe the physical and chemical properties of lipids, their synthesis and function in membranes and metabolism	2. Crit. Thinking 3. Prof. Competence
4. Describe the central catabolic and anabolic carbohydrate pathways (glycolysis, gluconeogenesis, pentose phosphate pathway, citric acid cycle, electron transport and ATP Synthesis) and carbohydrate metabolism	2. Crit. Thinking 3. Prof. Competence
5. Relate the principles of thermodynamics and kinetics to energy utilization, biological organization and metabolic pathways	2. Crit. Thinking 3. Prof. Competence
6. Describe the physical and chemical properties of nucleic acids, DNA and RNA; explain the process of inheritance and protein synthesis and how it relates to viruses, cancer and immunology	2. Crit. Thinking 3. Prof. Competence

K. TEXTS:

Campbell, Mary K., Farrell, Shawn O. *Biochemistry, 8<sup>th</sup> edition*, Cengage Learning, Stamford, CT 2012

L. REFERENCES:

McKee, T., and McKee, J.R., *Biochemistry, The Molecular Basis of Life, 3<sup>rd</sup> edition*, McGraw Hill, New York, NY 2003.

M. EQUIPMENT: Scientific calculator that can do logarithms and powers of 10

N. GRADING METHOD: A-F

O. MEASUREMENT CRITERIA:

Exams and Quizzes

Assignments

Final Exam

P. DETAILED COURSE OUTLINE:

7. Introduction to Biochemistry: Biochemistry and the Organization of Cells
  - a. Prokaryotic and Eukaryotic Cells
  - b. Thermodynamics and free energy
    - i. Role of ATP
8. Water: the Solvent for Biochemical Reactions
  - c. Properties of water and intermolecular forces
  - d. Acid base properties and buffers
9. Amino Acids and Peptides
  - e. Structure and properties of amino acids
  - f. The peptide bond and small peptides
10. The Three-Dimensional Structure of Proteins
  - g. Protein structure and Function
  - h. Primary structure
  - i. Secondary structure
  - j. Tertiary Structure (Globular and Fibrous structure)
    - i. Myoglobin
  - k. Quaternary structures of proteins
    - i. Hemoglobin
  - l. Protein denaturation and folding
11. Protein Purification and Characterization Techniques
12. The Behavior of Proteins: Enzymes
  - m. enzymes as catalysts
  - n. enzyme kinetics

- o. enzyme substrate binding
- p. enzyme inhibition (competitive, non-competitive)
- 13. The Behavior of Proteins: Enzymes, Mechanisms, and Control
  - q. Allosteric enzymes
  - r. Phosphorylation
  - s. Active Sites
  - t. Zymogens, Coenzymes and Vitamins
- 14. Lipids and Membranes
  - u. The chemical Nature of the lipid types
    - i. Fatty acids, triglycerides, phosphoglycerides, waxes, sphingolipids, glycolipids and steroids
  - v. Biological membranes
    - i. Lipid bilayers
    - ii. Membrane proteins and Fluid mosaic model
    - iii. Function of membranes
    - iv. Lipid soluble vitamins, prostaglandins and leukotrienes
- 15. Nucleic Acids: How Structure Conveys Information
  - w. Structures and properties of nucleotides and nucleic acids
    - i. How DNA and RNA differ
  - x. Structure of DNA
    - i. Chromosomes and chromatin
  - y. Denaturation of DNA
  - z. Principle Kinds of RNA and their Structure
- 16. Biosynthesis of Nucleic Acids: Replication
  - aa. DNA Replication and polymerase
    - i. Polymerization reaction mechanisms
  - bb. Repair and Recombination
    - i. mutation
- 17. Transcription of the Genetic Code: Biosynthesis of RNA
  - cc. Transcription in prokaryotes and eukaryotes
    - i. RNA Processing
  - dd. Ribosomes
- 18. Protein Synthesis: Translation of the Genetic Message
  - ee. The genetic code: Gene expression
  - ff. Codon-anticodon interaction
  - gg. Amino acid activation
  - hh. Prokaryotic and Eukaryotic Translation
  - ii. Protein degradation
- 19. Nucleic Acids Biotechnology Techniques
  - jj. Genetic engineering
    - i. Recombinant DNA Technology and cloning
  - kk. DNA libraries and Fingerprinting
- 20. Viruses, Cancer and Immunology
  - ll. Viruses and retroviruses
  - mm. The immune system, cancer and aids
- 21. Electron Transfer in Metabolism

- 22. Carbohydrates
  - nn. Structures and properties of mono-, di- and polysaccharides, glycoconjugates
- 23. Glycolysis
  - oo. (reactions of the glycolic pathway)
- 24. Gluconeogenesis,
  - pp. Pentose Phosphate Pathway and Glycogen Metabolism
- 25. The Citric Acid Cycle
  - qq. Oxidation reduction reactions
- 26. Electron Transport and ATP Synthesis
  - rr. Electron transport components and inhibitors
  - ss. Oxidative phosphorylation and chemiosmotic coupling
- 27. Lipid Metabolism
  - tt. Fatty acid degradation
- 28. Photosynthesis
- 29. The Metabolism of Nitrogen
  - uu. Nitrogen fixation
  - vv. Amino Acid Catabolism
    - i. Urea cycle